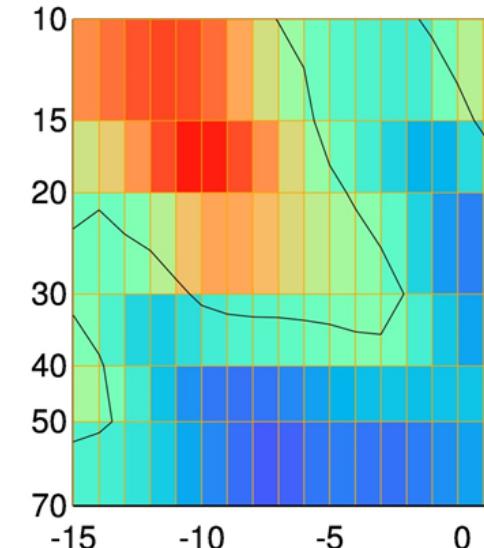
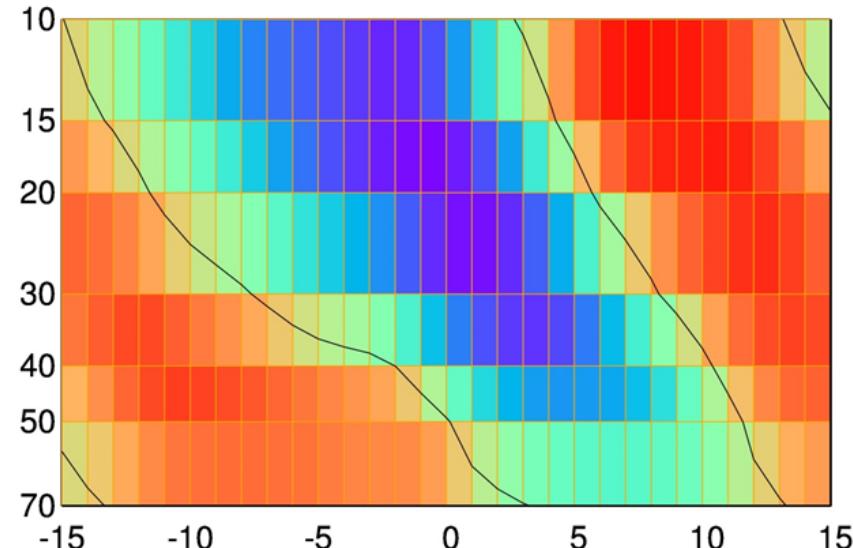
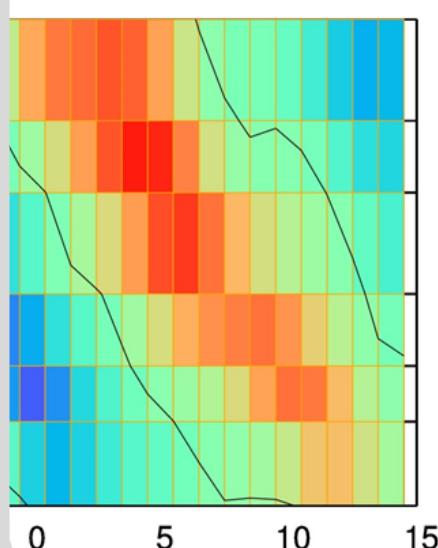


Chemistry-Climate Modelling

Peter Braesicke

IMK-ASF



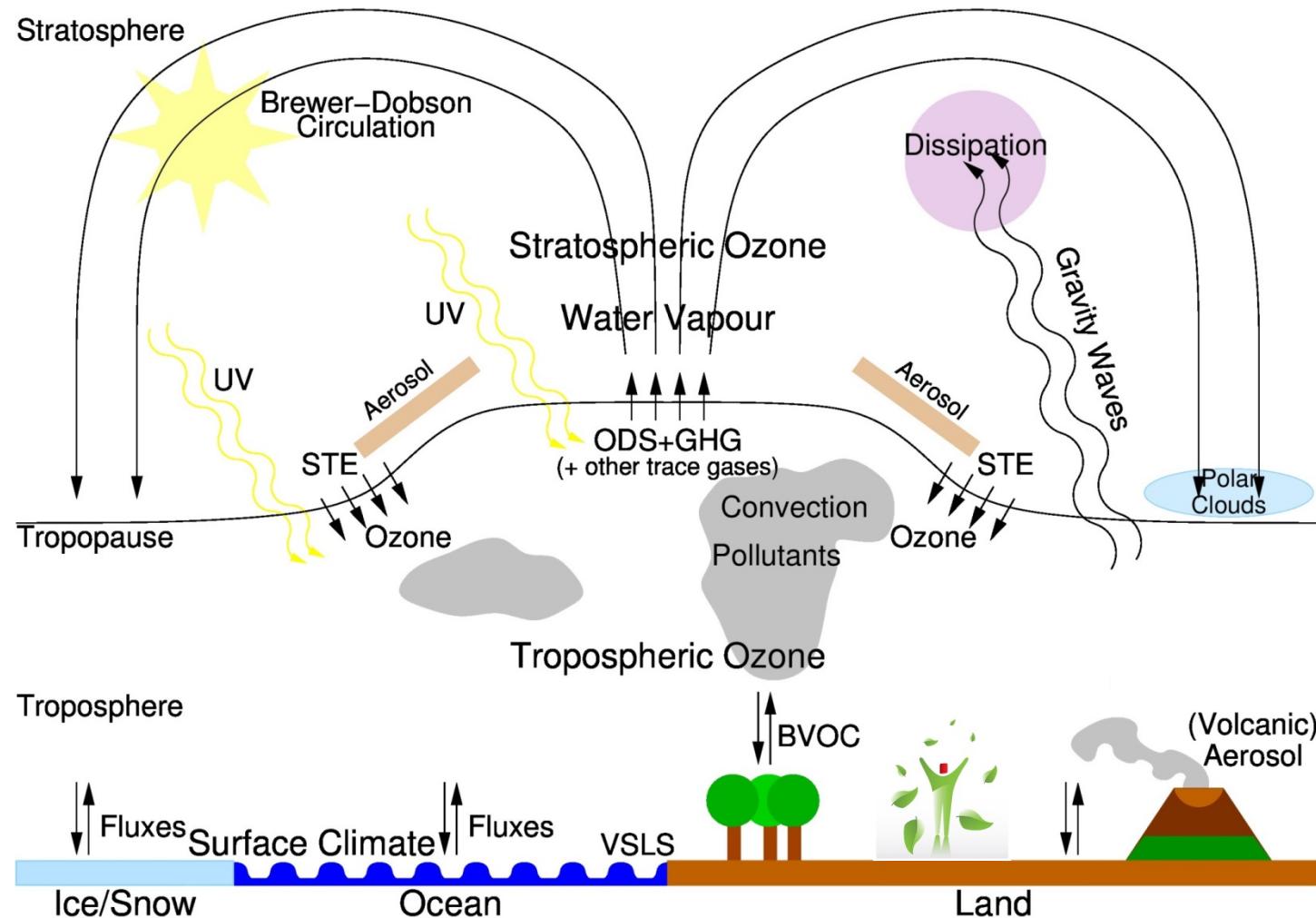
Structure

- What is a Chemistry-Climate Model?
 - Model components (sub-models)
 - Model validation (confrontation with observations)
 - CCMs in the context of a model hierarchy
- Examples of processes (Practical, [BDC]):
 - High latitudes: Polar Ozone Depletion
 - [Low latitudes: Quasi-Biennial Oscillation]

What is a Chemistry-Climate Model?

MODEL COMPONENTS

A complex system ...



... and we are right in the middle!

Chemistry-Climate Model (CCM) Schematic

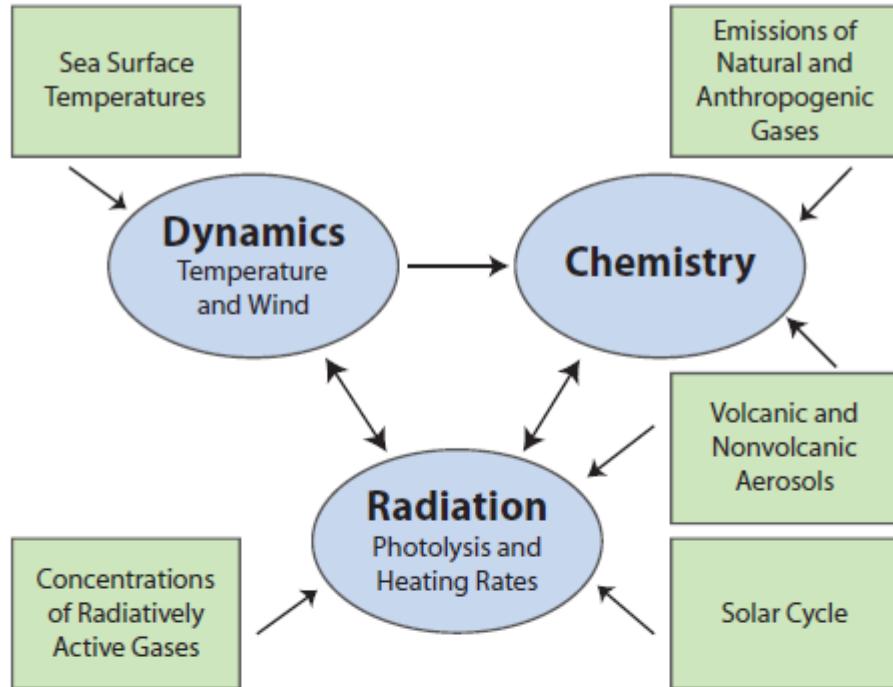


Figure 5-1. Schematic of a Chemistry-Climate Model (CCM). The core of a CCM (oval symbols) consists of an atmospheric general circulation model (AGCM) that includes calculation of the heating and cooling rates and a detailed chemistry module. They are interactively coupled. Photolysis rates are calculated online or are determined from a lookup table. Arrows indicate the direction of effect. Rectangular boxes denote external impacts. In current CCMs, sea surface temperatures (SSTs) are prescribed based on observations or are adopted from calculations with a climate model. Natural and anthropogenic emissions of gases are considered. Tropospheric and stratospheric aerosol loading (especially after volcanic eruptions) can be taken into account. CCMs often consider the changes of solar radiation caused by the 11-year activity cycle of the Sun.

Scientific Assessment of Ozone Depletion: 2006

Dynamics (Thermodynamics)

■ Example: Primitive Equations

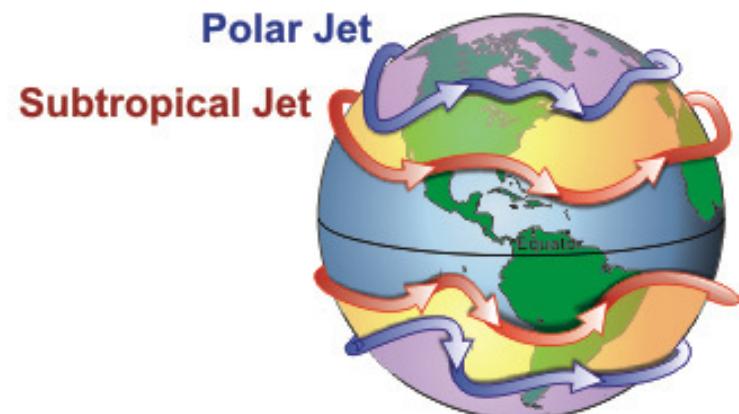
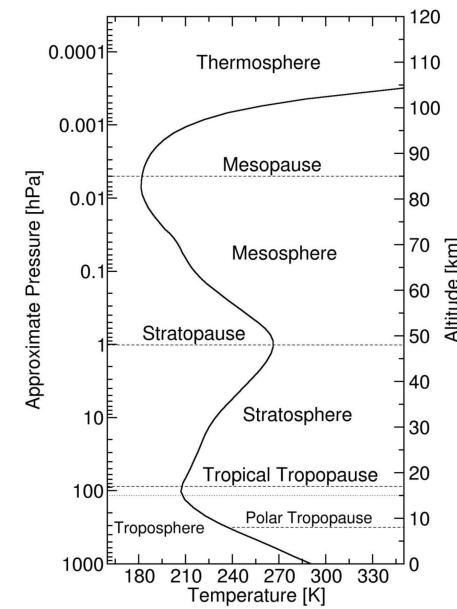
■ Equations of Motion

- $\partial u / \partial t = -v \cdot \nabla u - \omega \partial u / \partial p + fv - \partial \Phi / \partial x$
- $\partial v / \partial t = -v \cdot \nabla v - \omega \partial v / \partial p - fu - \partial \Phi / \partial y$

Vertical Coordinates:

↓ Pressure (p)

Height (z) ↑



Dynamics (Thermodynamics)

■ Example: Primitive Equations

■ Equations of Motion

- $\partial u / \partial t = -v \cdot \nabla u - \omega \partial u / \partial p + fv - \partial \Phi / \partial x$
- $\partial v / \partial t = -v \cdot \nabla v - \omega \partial v / \partial p - fu - \partial \Phi / \partial y$

■ Hydrostatic Assumption (Equation)

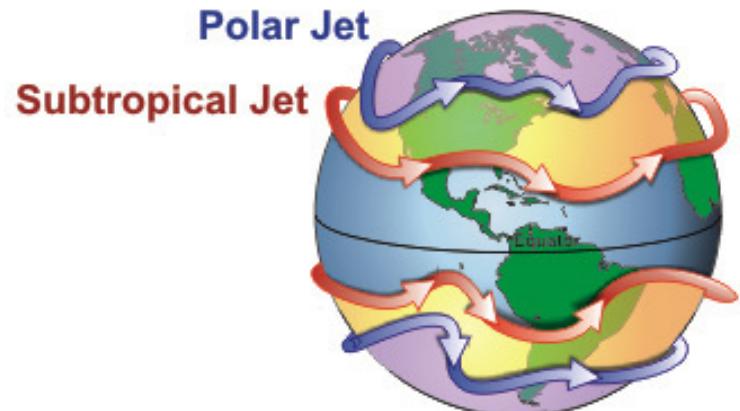
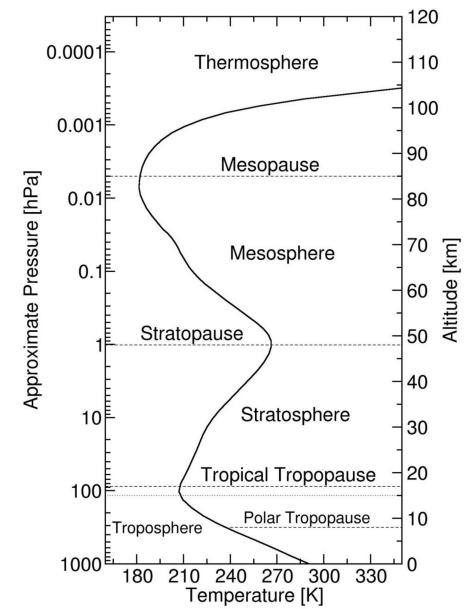
- $\partial \Phi / \partial p = -1 / \rho$

■ Temperature Tendency Equation

- $\partial T / \partial t = -v \cdot \nabla T - \omega \partial T / \partial p + \omega \kappa T / p + T / c \downarrow p \ ds / dt$

■ Continuity Equation

- $\partial \omega / \partial p = -\partial u / \partial x - \partial v / \partial y$



Dynamics (Thermodynamics)

■ Example: Primitive Equations

■ Equations of Motion

- $\partial u / \partial t = -v \cdot \nabla u - \omega \partial u / \partial p + fv - \partial \Phi / \partial x$
- $\partial v / \partial t = -v \cdot \nabla v - \omega \partial v / \partial p - fu - \partial \Phi / \partial y$

■ Hydrostatic Assumption (Equation)

- $\partial \Phi / \partial p = -1 / \rho$

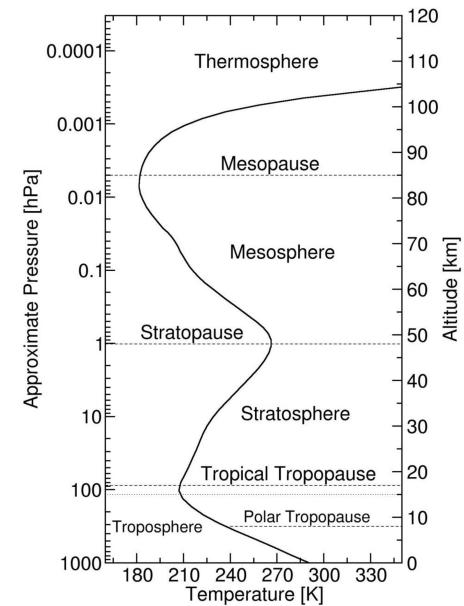
■ Temperature Tendency Equation

- $\partial T / \partial t = -v \cdot \nabla T - \omega \partial T / \partial p + \omega \kappa T / p + \frac{\partial \ln \rho}{\partial p} \frac{\partial \ln \rho}{\partial t}$

Vertical Coordinates:

↓ Pressure (p)

Height (z) ↑

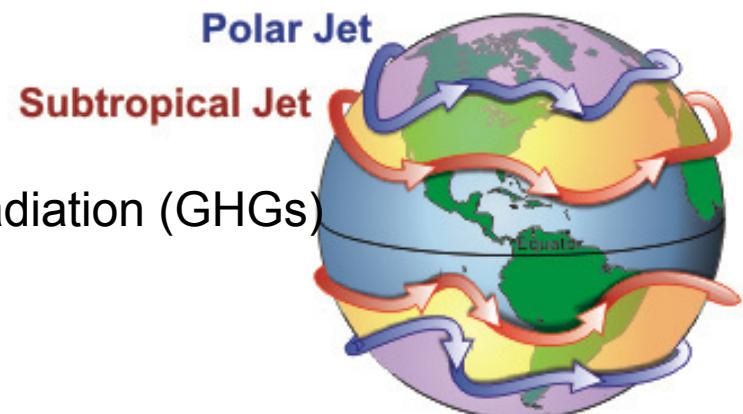


■ Continuity Equation

- $\partial \omega / \partial p = -\partial u / \partial x - \partial v / \partial y$

■ 5 Variables: (u, v, w, T, Φ) with $\rho = \rho(T, p)$

■ Missing: Specific Humidity (H_2O) and Radiation (GHGs)



Dynamics (Background)

■ Geopotential:

- $g = d\Phi/dz$; $d\Phi = gdz$
- $[\Phi] = m^{12} / s^{12}$

■ Geopotential Height:

- $z = \Phi/g \downarrow 0$

■ Hydrostatic Equilibrium:

- $dp/dz = -\rho g$
- $gdz/dp = d\Phi/dp = -1/\rho$

■ Ideal Gas Law:

- $p = \rho RT$
- $(p = p \downarrow A + p \downarrow W)$

■ Vertical Velocity:

- $dp = -\rho g dz$
- $\omega = -\rho g w$

■ Enthalpy (specific):

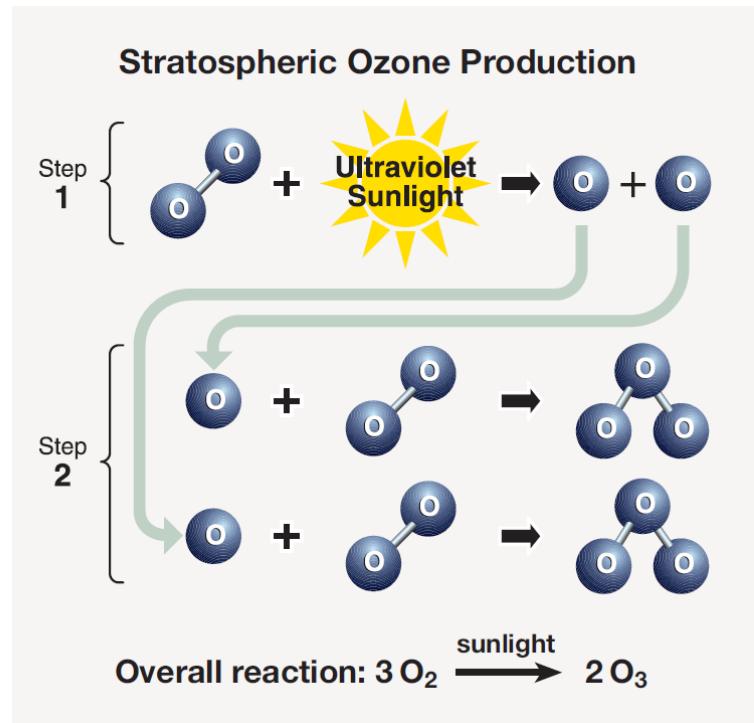
- $dh = 1/\rho dp + T ds$
- $dh = c \downarrow p dT$
- $\Rightarrow p = \text{const.} \rightarrow dT = T/c \downarrow p$
- ds

■ Heating Rate:

- $dT/dt = T/c \downarrow p$
- $ds/dt = R \downarrow A / c \downarrow p = 2/7$
- $[dT/dt] = K/s$
- $c \downarrow p = 1010 J/kgK$

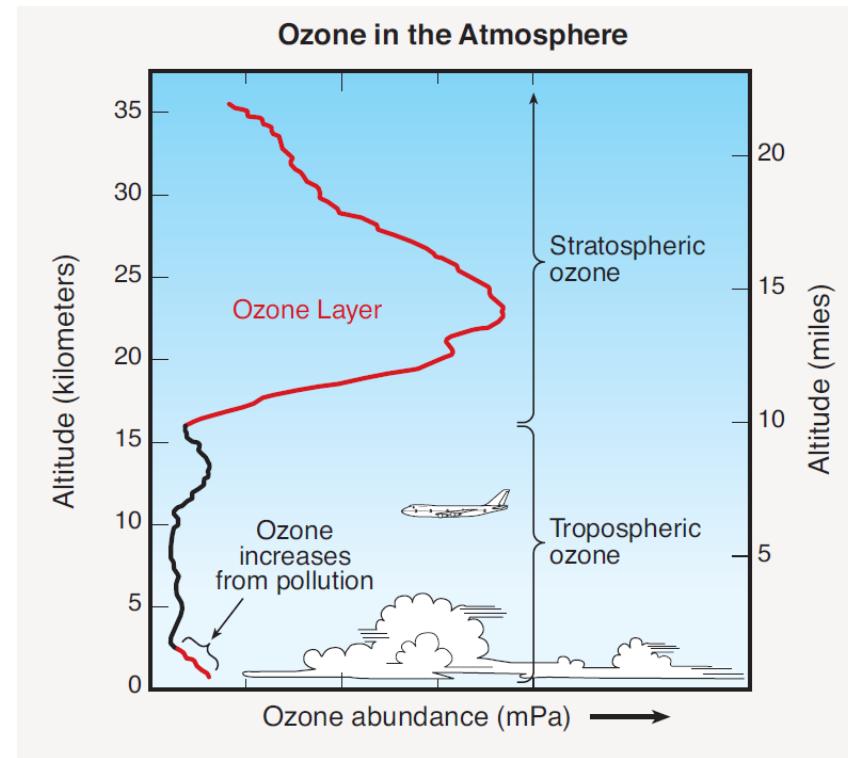
Chemistry (Radiation)

■ Gas Phase Ozone Chemistry



(Example)

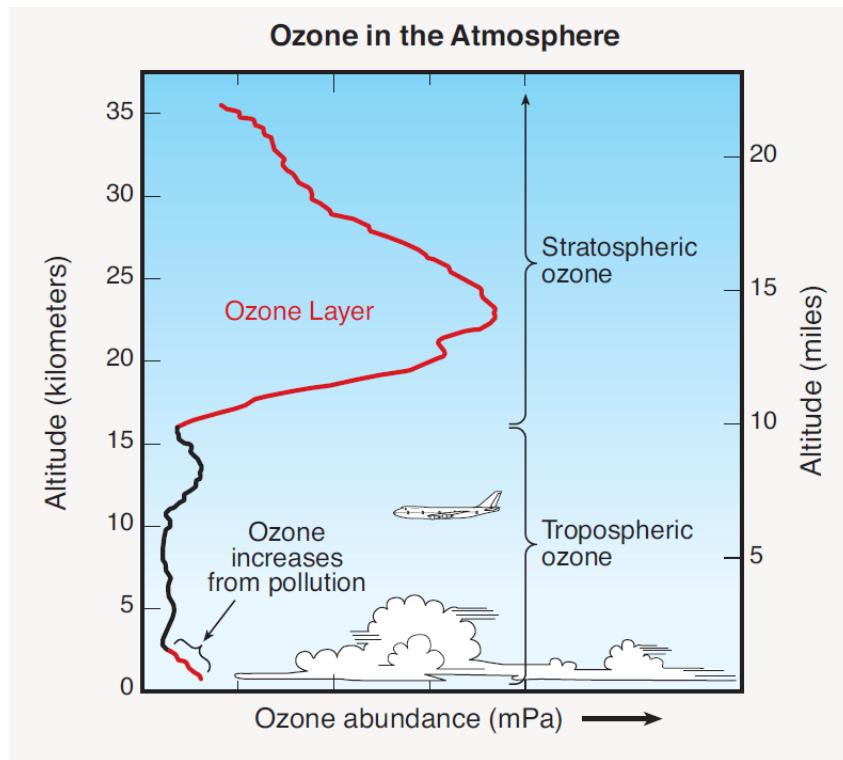
■ Vertical O₃ Distribution



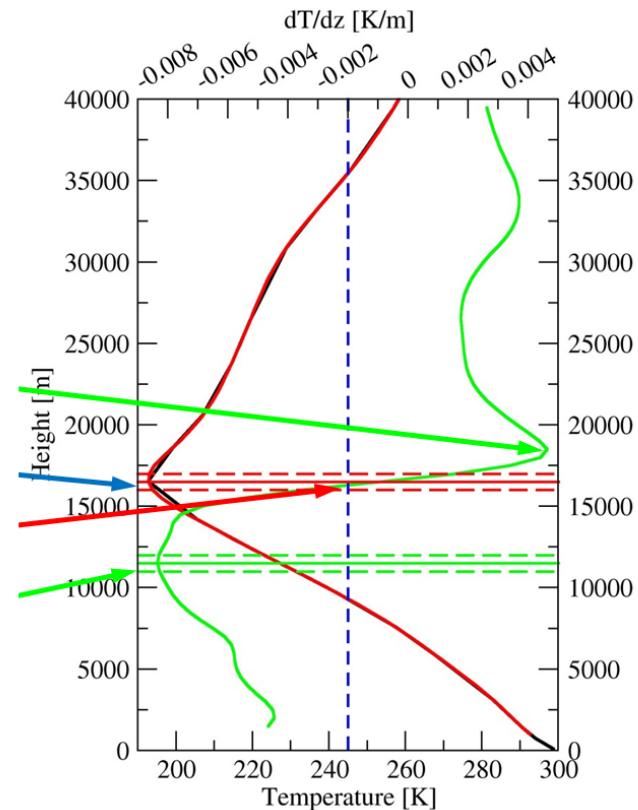
UNEP/WMO Ozone Assessment: 20 Questions

Chemistry (Radiation)

■ Vertical O₃ Distribution



■ Vertical Temperature Distribution



UNEP/WMO Ozone Assessment: 20 Questions

Chemistry (Radiation)

■ Vertical Temperature Distribution

Temperature (tropical latitudes)



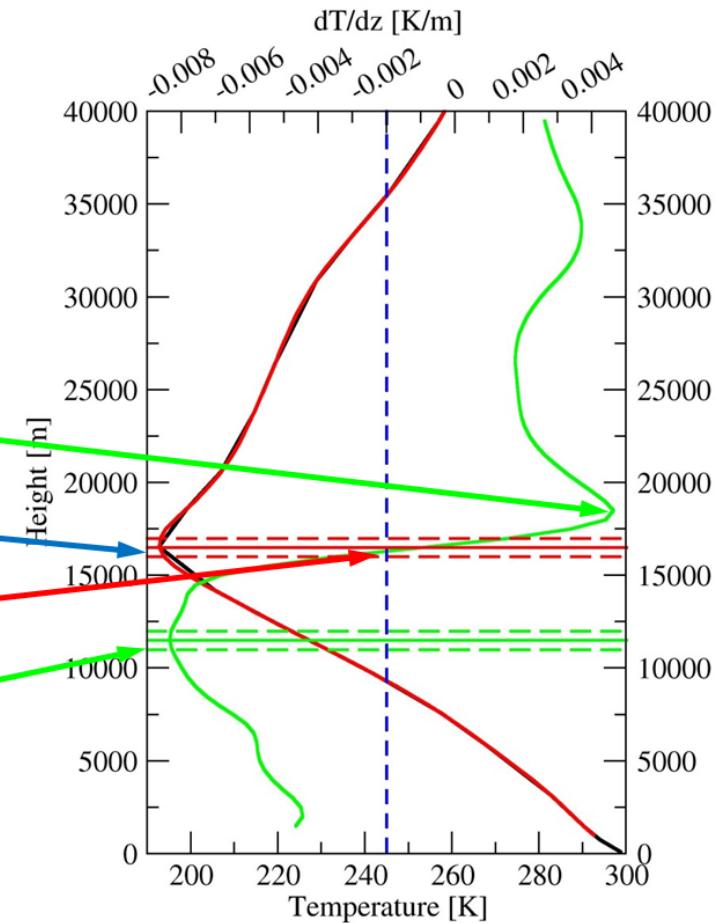
$$-\Gamma = \frac{dT}{dz} > -2 \frac{K}{km}$$

Lapse Rate Maximum

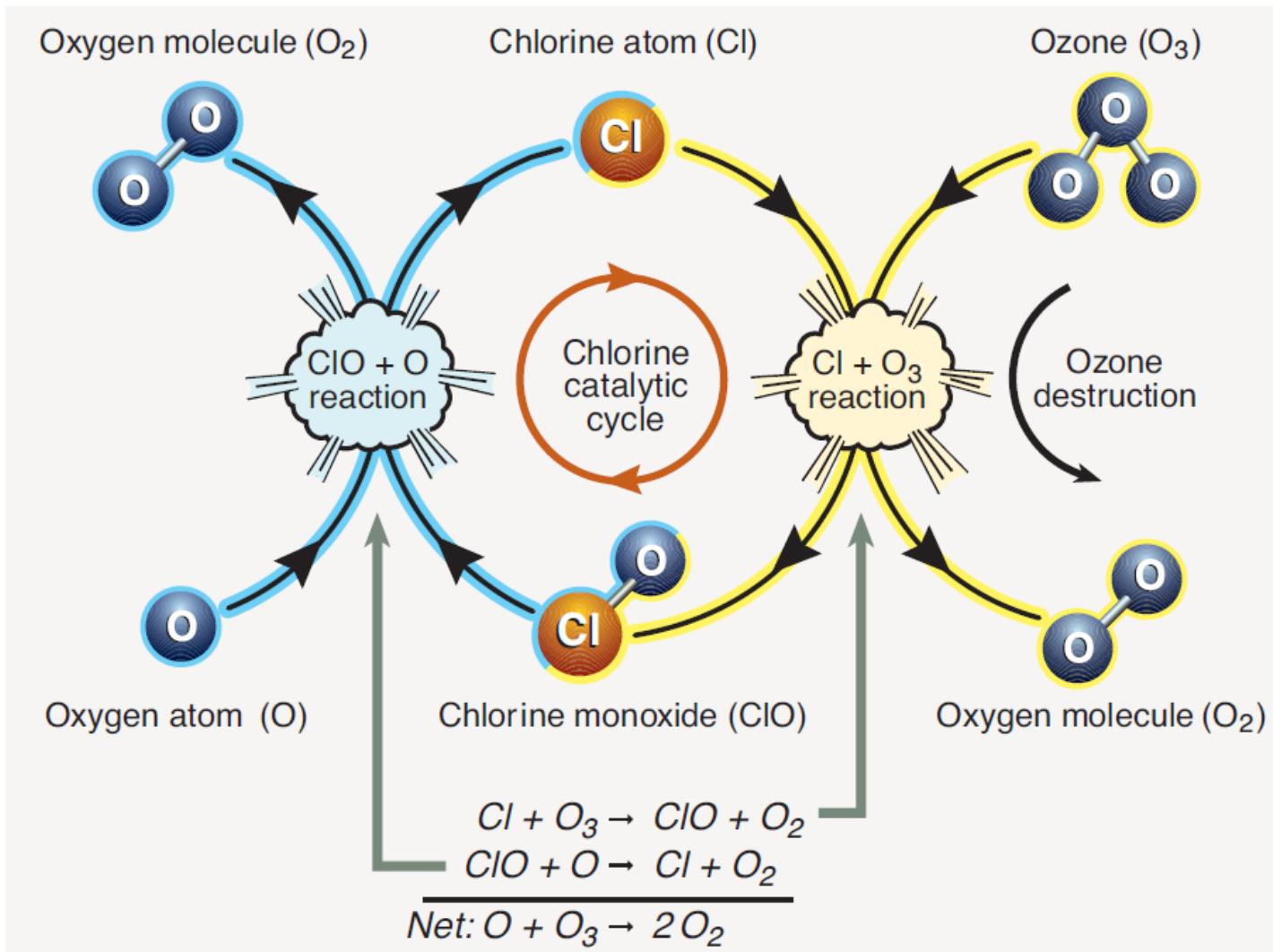
Cold Point

WMO Tropopause

Lapse Rate Minimum



More Chemistry: Ozone Destruction



UNEP/WMO Ozone Assessment: 20 Questions

Radiation (Thermodynamics)

■ Black Body / Full Insolation:

- $E_{out} = \sigma T^4$
- $E_{in} = 1/4 S_{\odot} = 342 W/m^2$
- $E_{in} = E_{out} \Rightarrow T_{\oplus} - T_{Earth} = 5^\circ C$

■ Black Body / Albedo Effect ($\alpha=0.3$)

- $E_{out} = \sigma T^4$
- $E_{in} = 1/4 (1-\alpha) S_{\odot} = 239 W/m^2$
- $E_{in} = E_{out} \Rightarrow T_{\oplus} - T_{Earth} = -18^\circ C$

■ GHG Effect ($\varepsilon=0.74$) / Albedo Effect

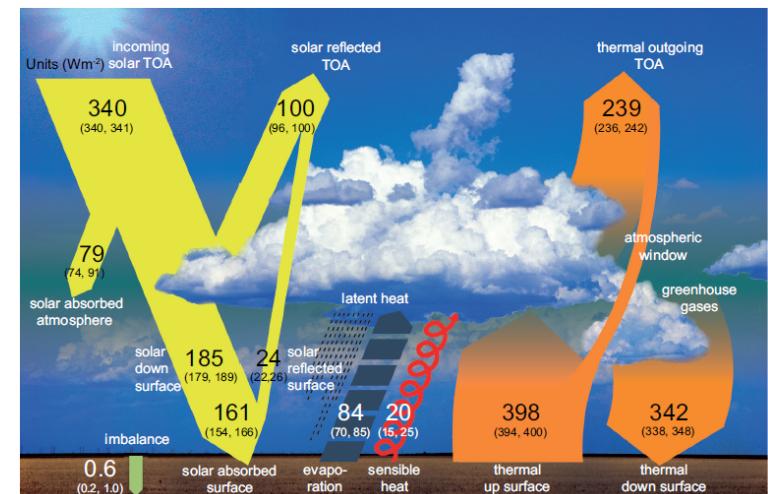
- $E_{out} = \varepsilon \sigma T^4$
- $E_{in} = 1/4 (1-\alpha) S_{\odot} = 239 W/m^2$
- $E_{in} = E_{out} \Rightarrow T_{\oplus} - T_{Earth} = 15^\circ C$

Fact:

$$T_{Earth} \sim 15^\circ C$$

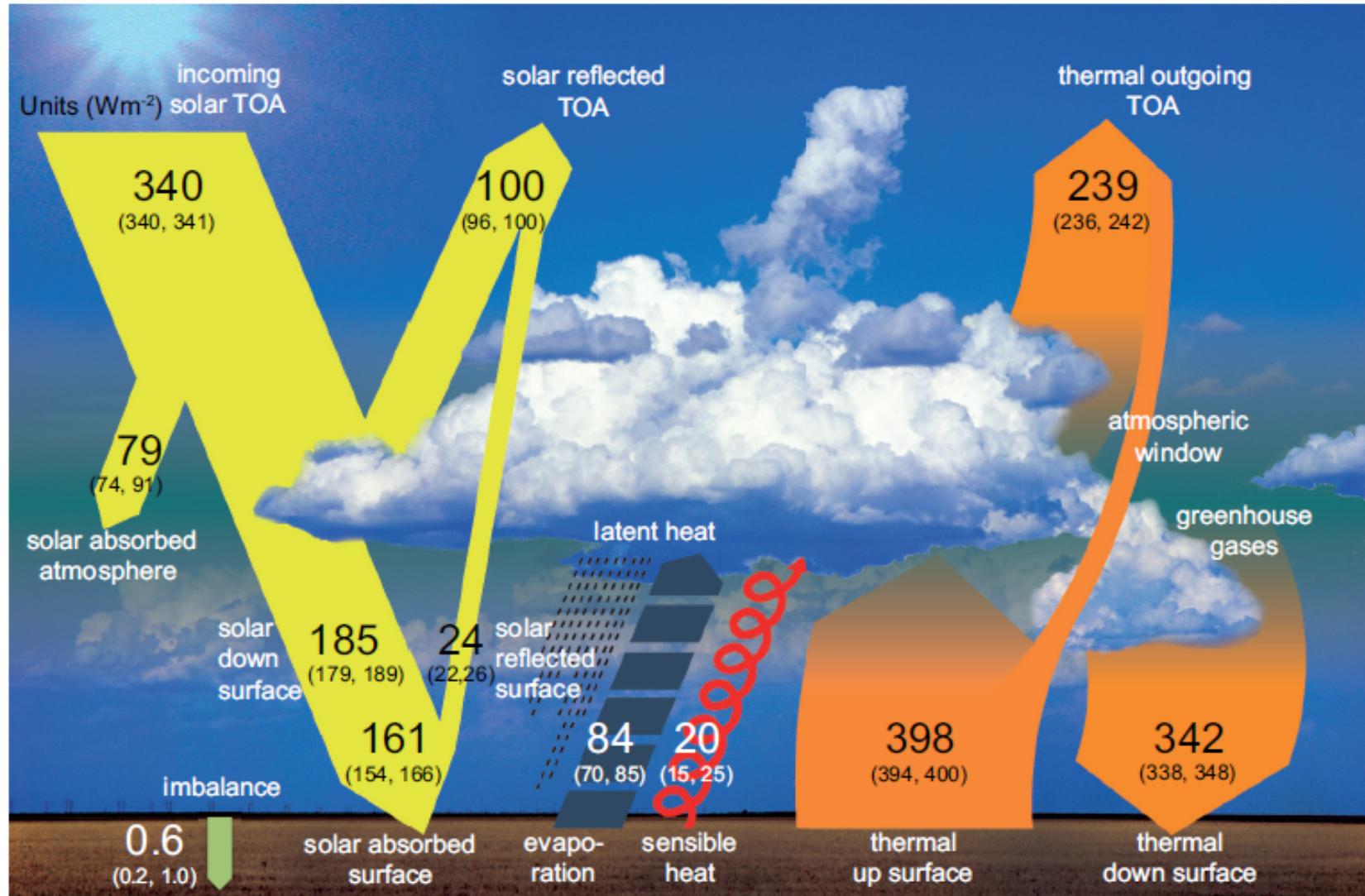
α :Albedo (~ 0.3)

ε :Emissivity (~ 0.74)

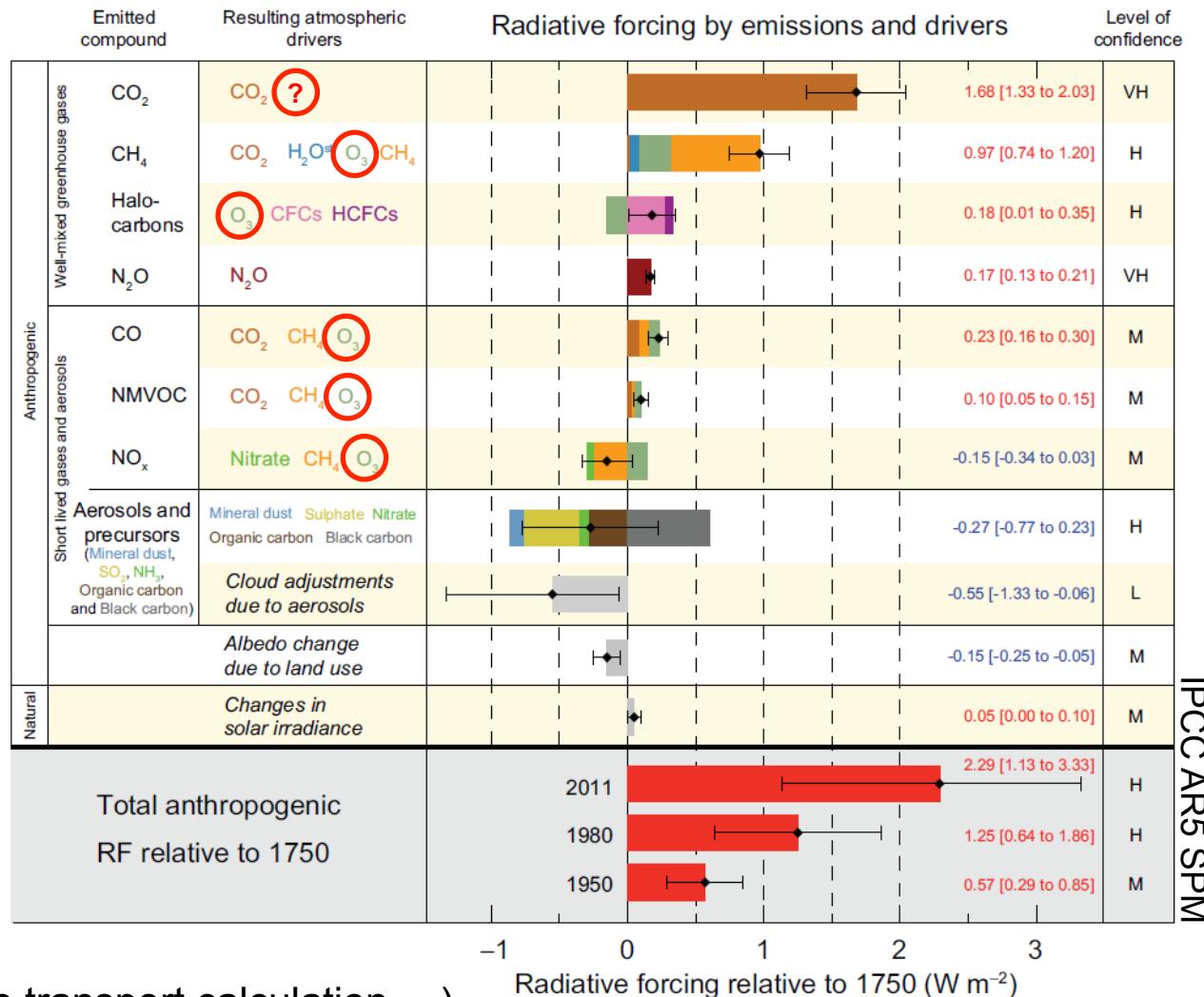


Radiation Budget

IPCC AR5

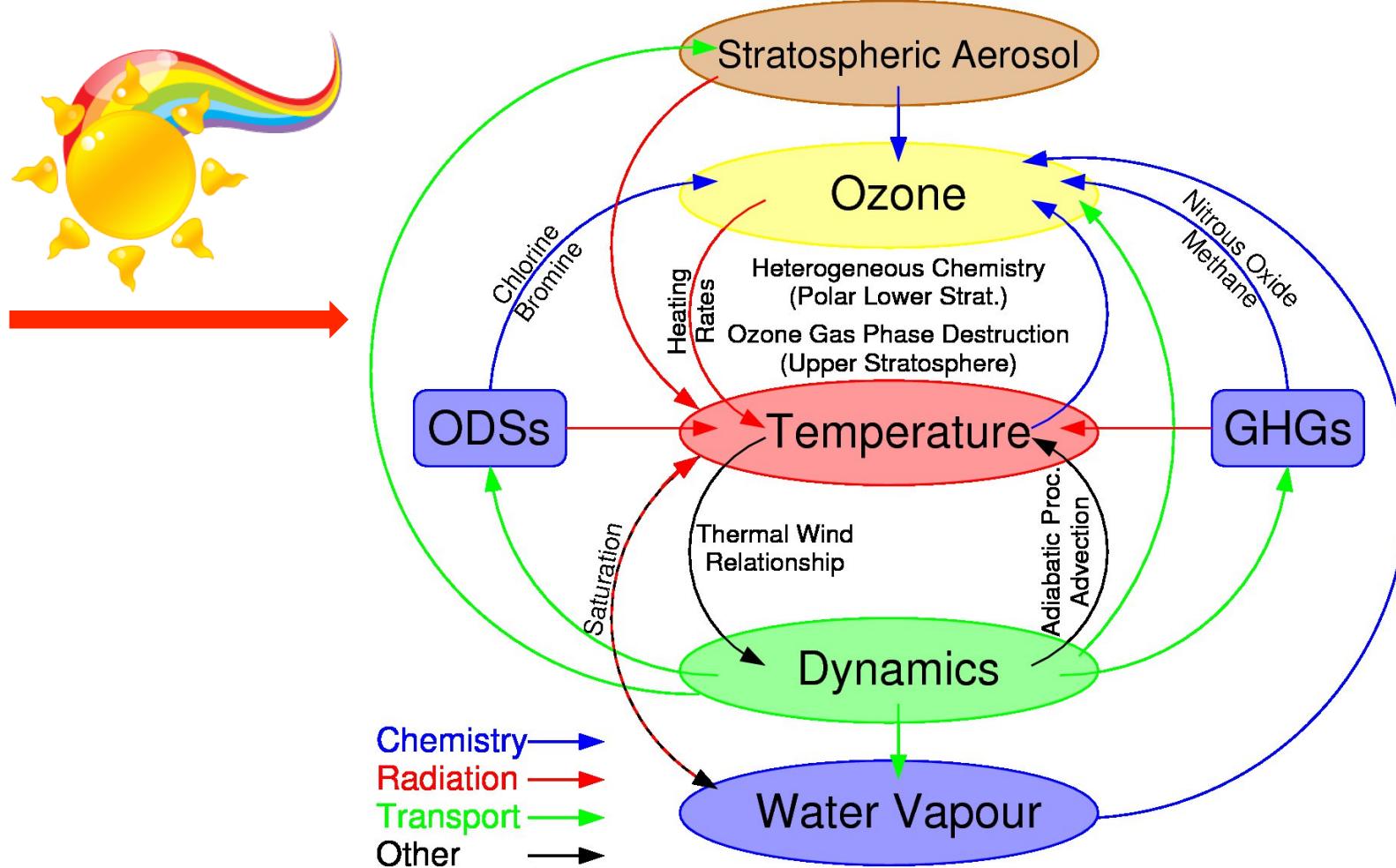


Radiative Forcing



(Full radiative transport calculation ...)

Interactions



Chemistry-Climate Model (CCM) Schematic

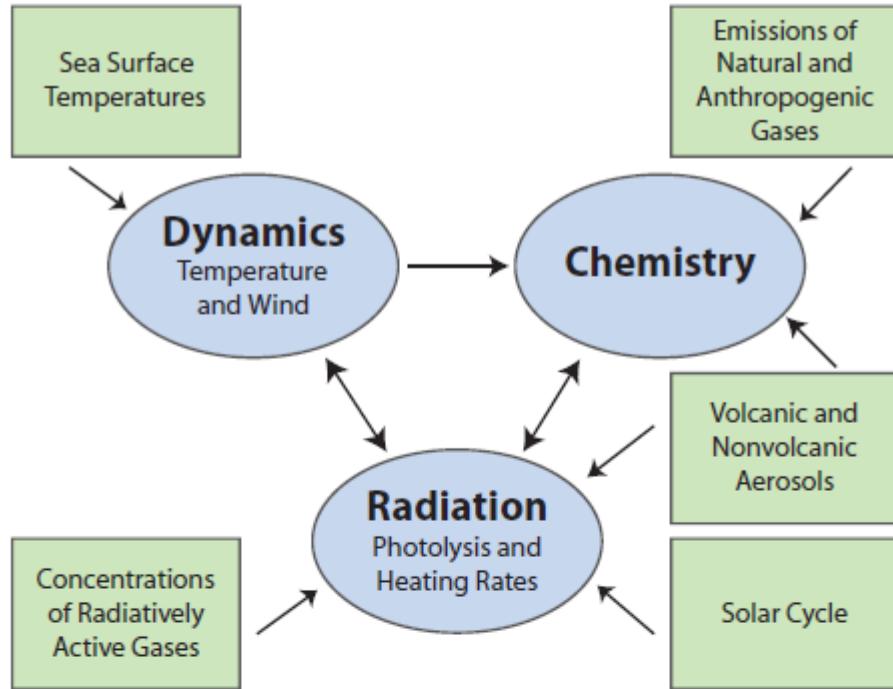


Figure 5-1. Schematic of a Chemistry-Climate Model (CCM). The core of a CCM (oval symbols) consists of an atmospheric general circulation model (AGCM) that includes calculation of the heating and cooling rates and a detailed chemistry module. They are interactively coupled. Photolysis rates are calculated online or are determined from a lookup table. Arrows indicate the direction of effect. Rectangular boxes denote external impacts. In current CCMs, sea surface temperatures (SSTs) are prescribed based on observations or are adopted from calculations with a climate model. Natural and anthropogenic emissions of gases are considered. Tropospheric and stratospheric aerosol loading (especially after volcanic eruptions) can be taken into account. CCMs often consider the changes of solar radiation caused by the 11-year activity cycle of the Sun.

Missing: Boundary conditions, numerical treatment

Scientific Assessment of Ozone Depletion: 2006

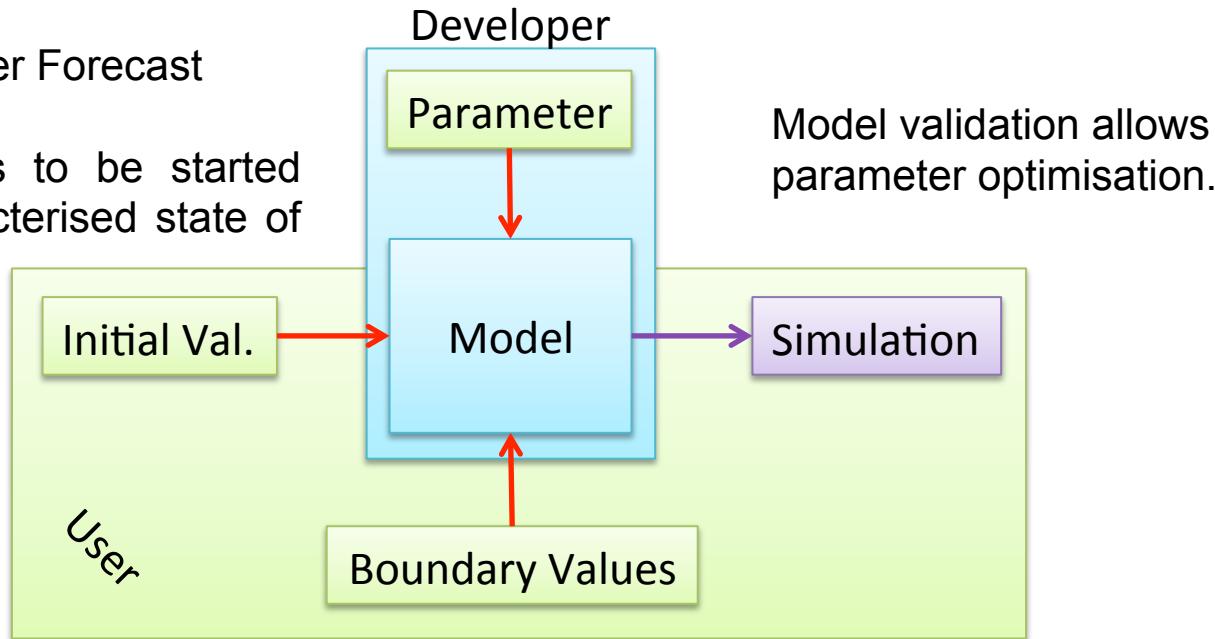
What is a Chemistry-Climate Model?

MODEL VALIDATION

Model and Auxiliary Data

Initial Values: (Chemical) Weather Forecast

The model needs to be started from a well characterised state of the atmosphere.

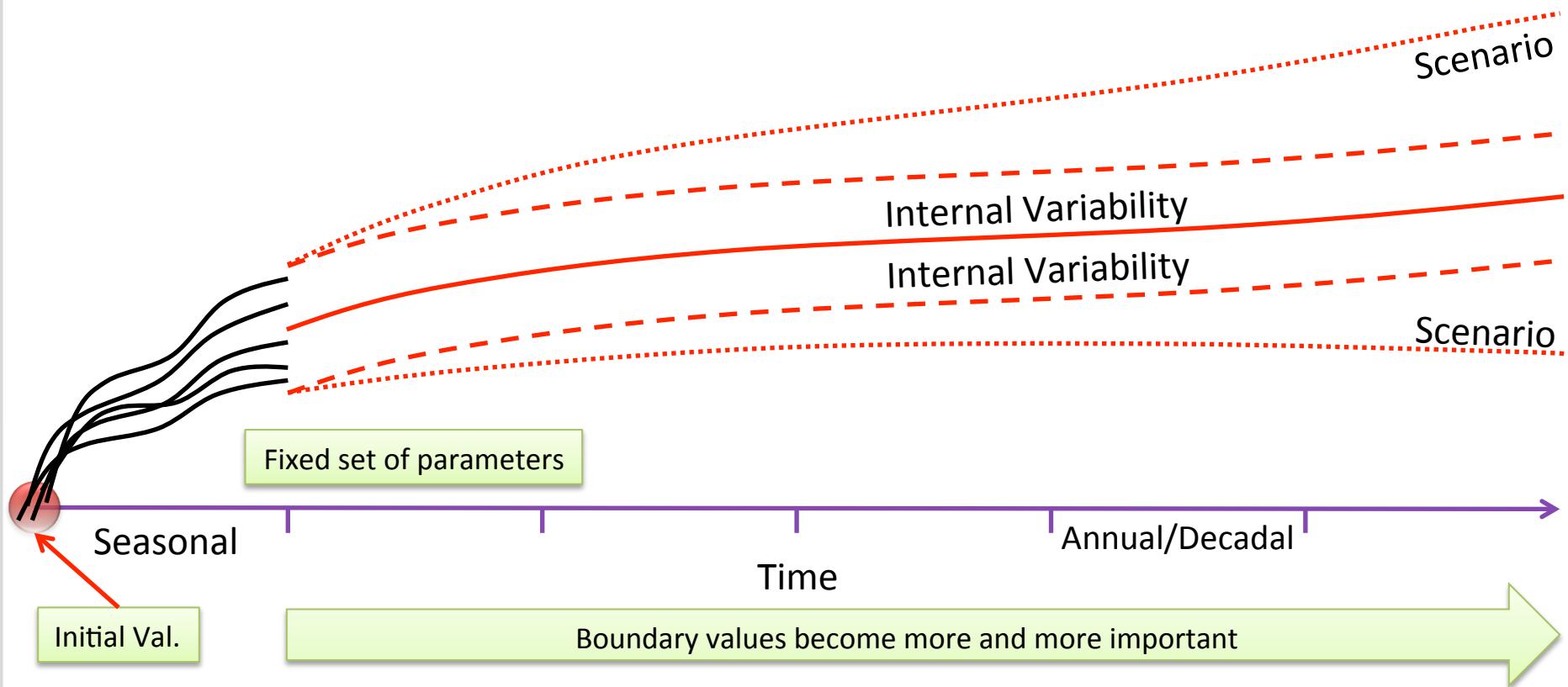


Model validation allows parameter optimisation.

Boundary Values: (Chemical) Climate Prediction

The model needs to know about expected changes (scenarios). How much CO₂ is emitted due to the use of fossil fuels?

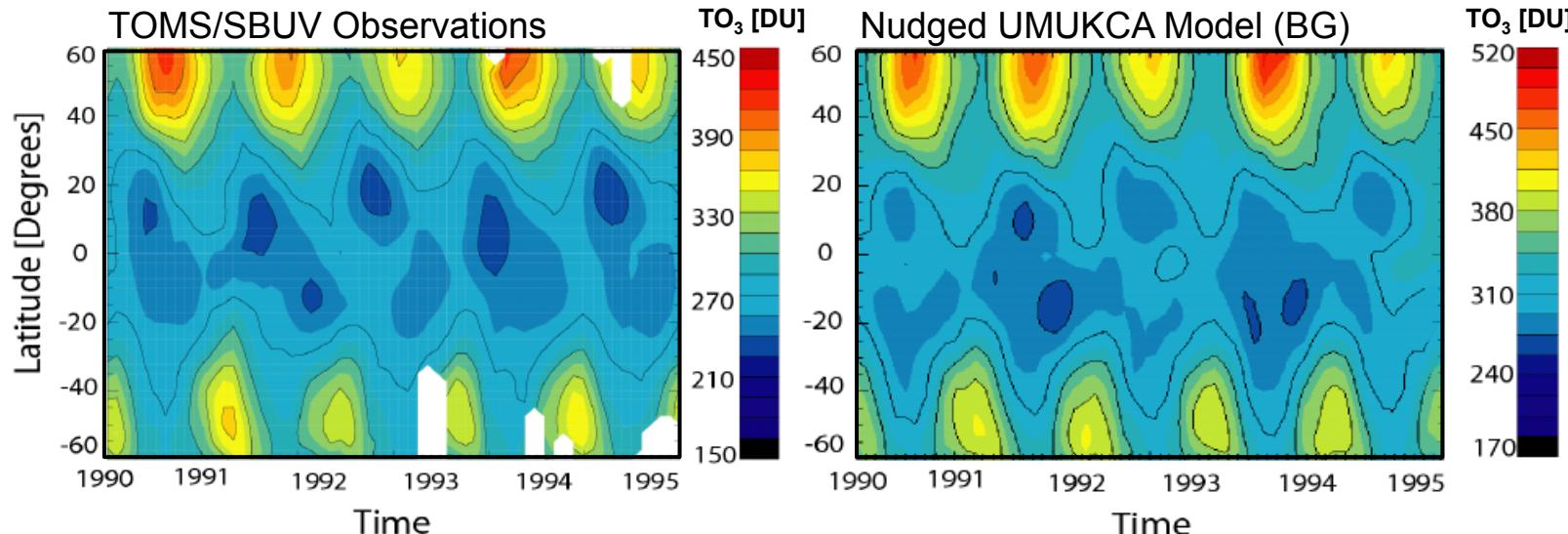
Predicting Weather and Climate



Model Validation

- Two Approaches:
 - Define a constraint in the model, so that some results of the model become directly comparable with observations, e.g. nudge meteorology to compare trace gases.
 - Simulate a particular case (year) over and over again and compare the statistics of the time-slice results to the measurements and their error bars.

Nudged Model Validation / Attribution



There are biases, but we look at differences!

Attribution:

Chemical effect of volcanic aerosol on ozone:

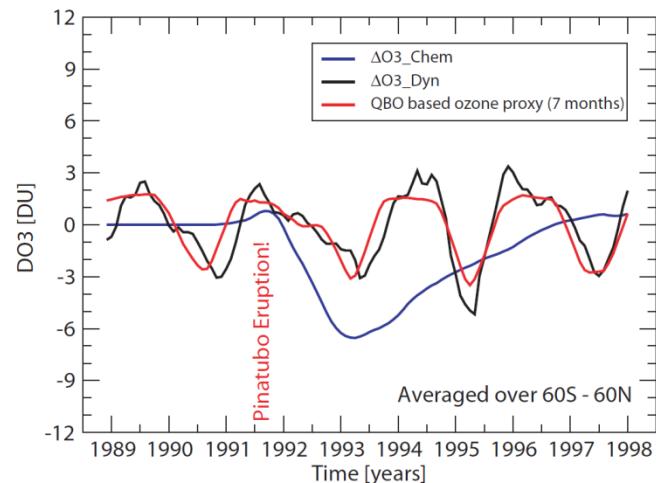
The model can run with and without the Pinatubo eruption.

Dynamical changes in ozone (check):

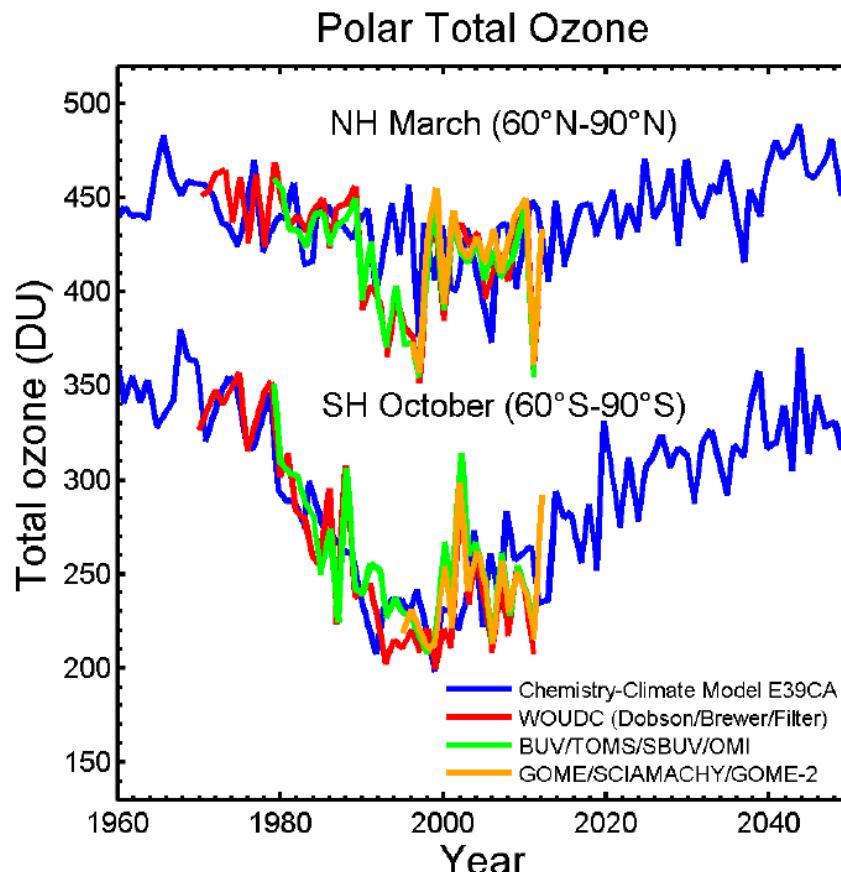
Observations-(chemical effect);

the residuum can be explained by dynamics

Telford, P., Braesicke, P., Morgenstern, O., and Pyle, J.: Reassessment of causes of ozone column variability following the eruption of Mount Pinatubo using a nudged CCM, *Atmos. Chem. Phys.*, 9, 4251-4260, doi:10.5194/acp-9-4251-2009, 2009.



Transient Changes



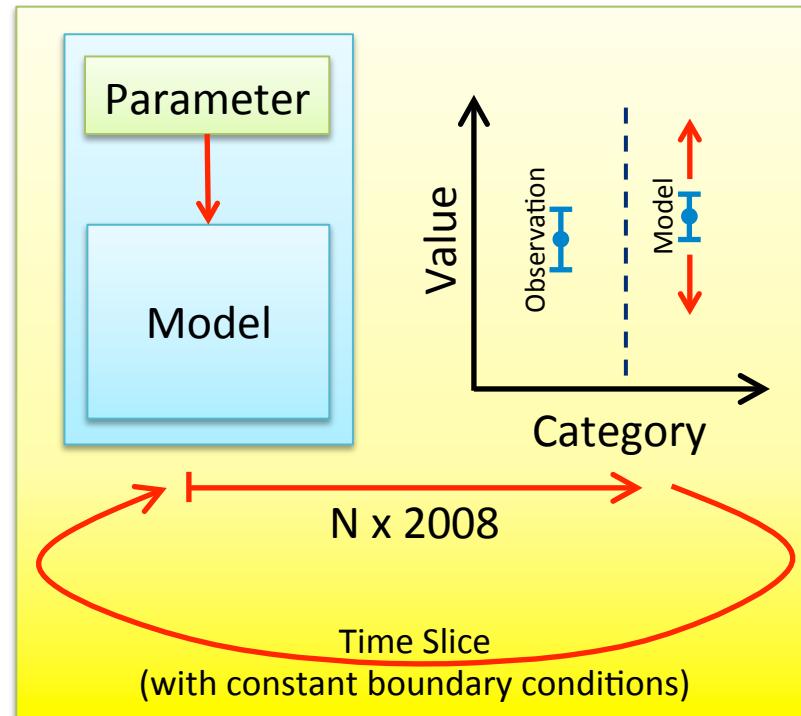
Total polar ozone in the northern and southern hemispheres as measured by various instruments, including GOME, SCIAMACHY and GOME-2 flown on ERS-2, Envisat and MetOp, respectively (in orange). The blue line depicts projections based on simulations from the Chemistry Climate Model E39CA. The total ozone reached its lowest levels in both hemispheres in the late 1990s, and it is expected to increase in the coming years.



Credits:
ESA/DLR/Eumetsat/NASA/WMO
/GAW

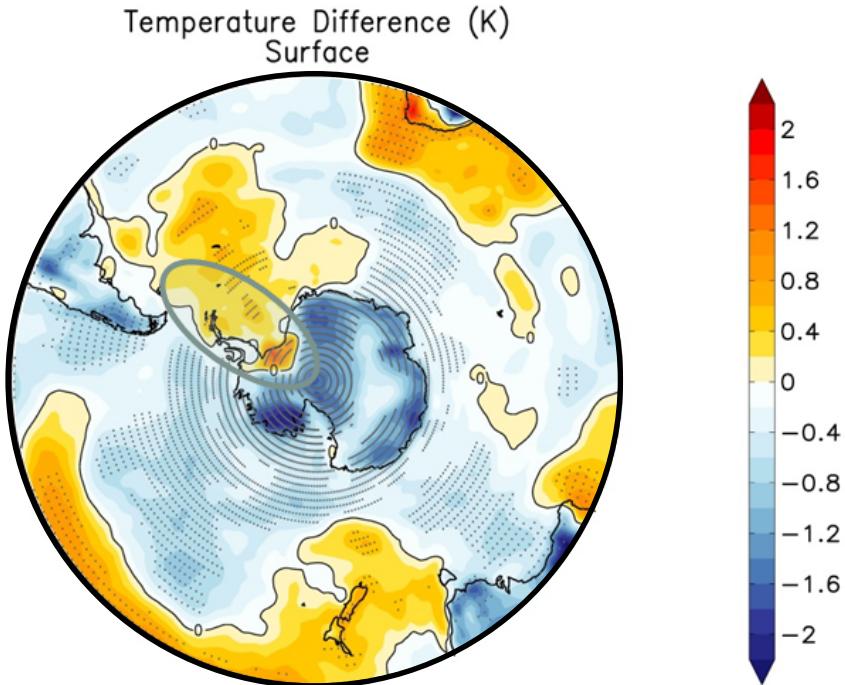
Dameris, M. and D. Loyola, Recent and future evolution of the stratospheric ozone layer, Chapter 45 in Atmospheric Physics, Background-Methods-Trends, Ed. U. Schumann, Springer Heidelberg New York Dordrecht London, doi: 10.1007/978-3-642-30183-4, ISBN 978-3-642-30182-7, pp. 747-761, 2012.

Statistical Model Validation



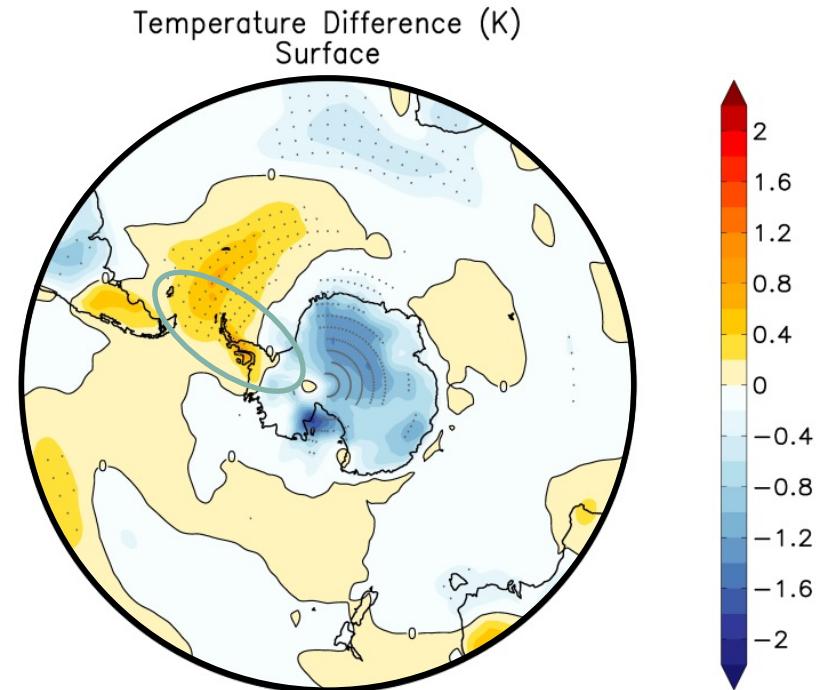
Antarctic Climate Change

„Observations“



ERA-Interim Reality:
Mean(1998 to 2002) - Mean(1979 to 1983)

Chemistry-Climate Model

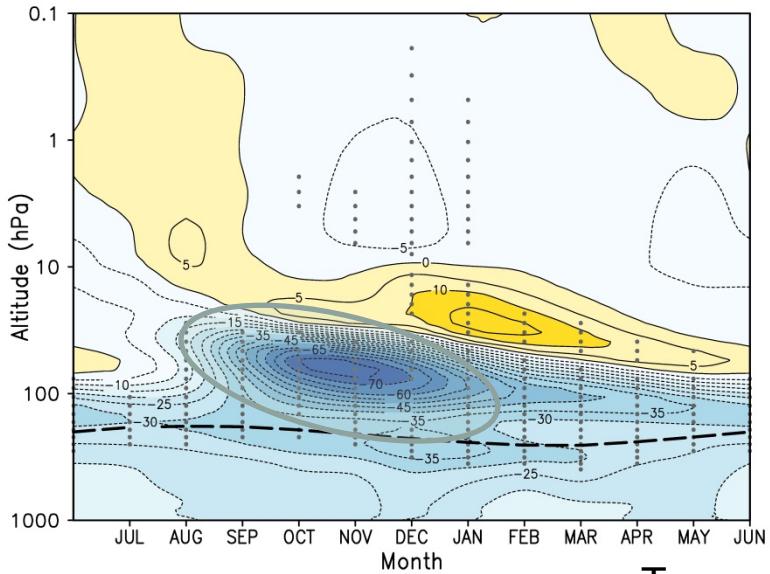


UM-UKCA:
Ozone Hole (~2000) – No OH (~2000)

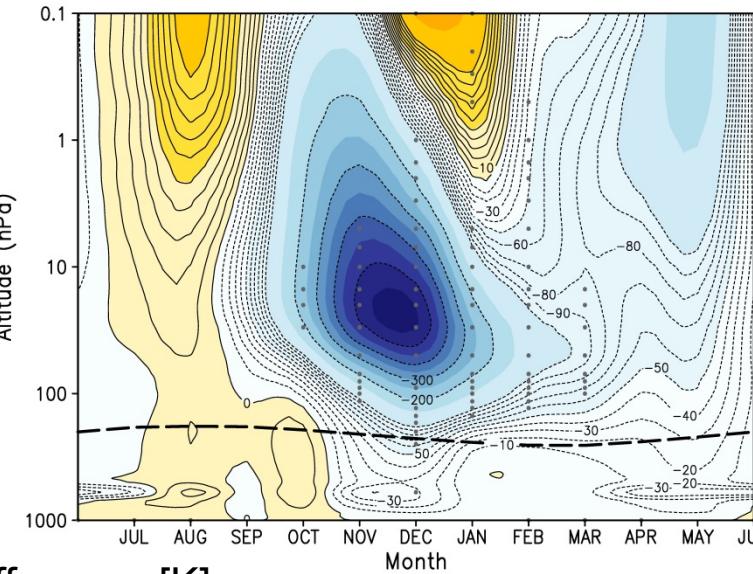
Keeble, J., Braesicke, P., Abraham, N. L., Roscoe, H. K., and Pyle, J. A.: The impact of polar stratospheric ozone loss on Southern Hemisphere stratospheric circulation and climate, *Atmos. Chem. Phys. Discuss.*, 14, 18049-18082, doi:10.5194/acpd-14-18049-2014, 2014.

Ozone Hole and Climate Change Attribution

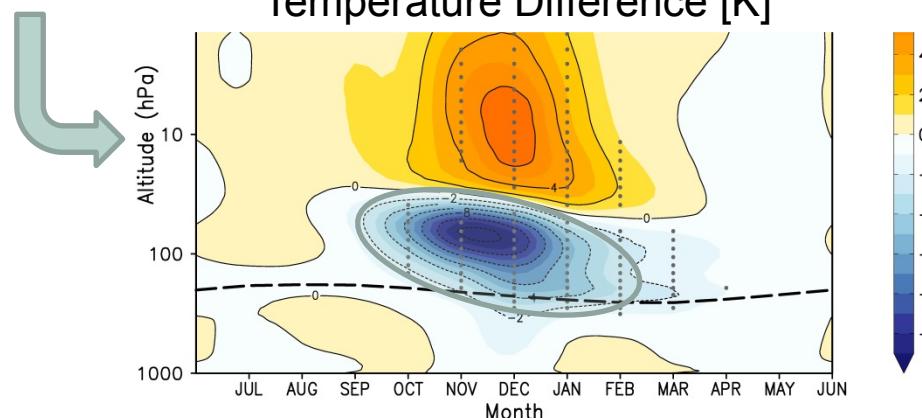
Ozone Difference [%] – 90°-75°S



Height Difference [m] – 90°-75°S



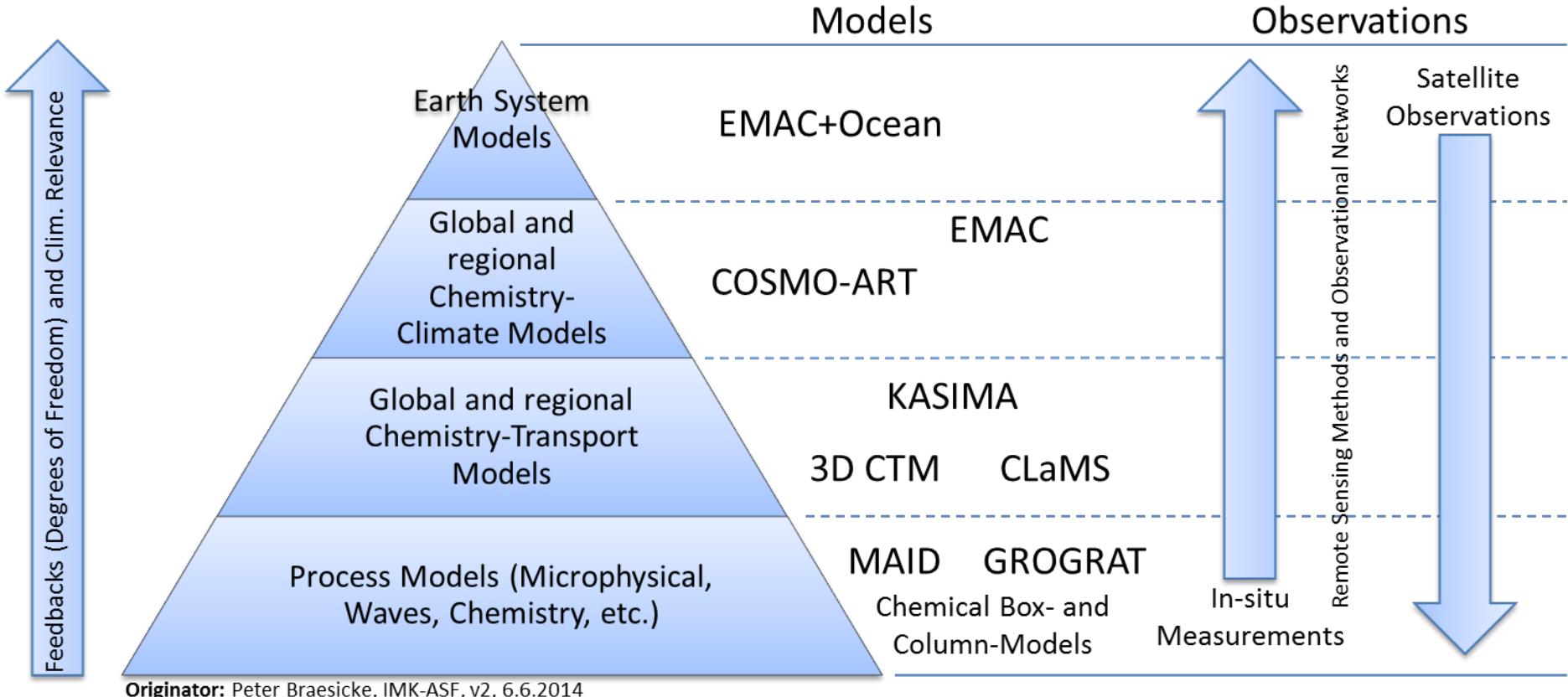
Temperature Difference [K]



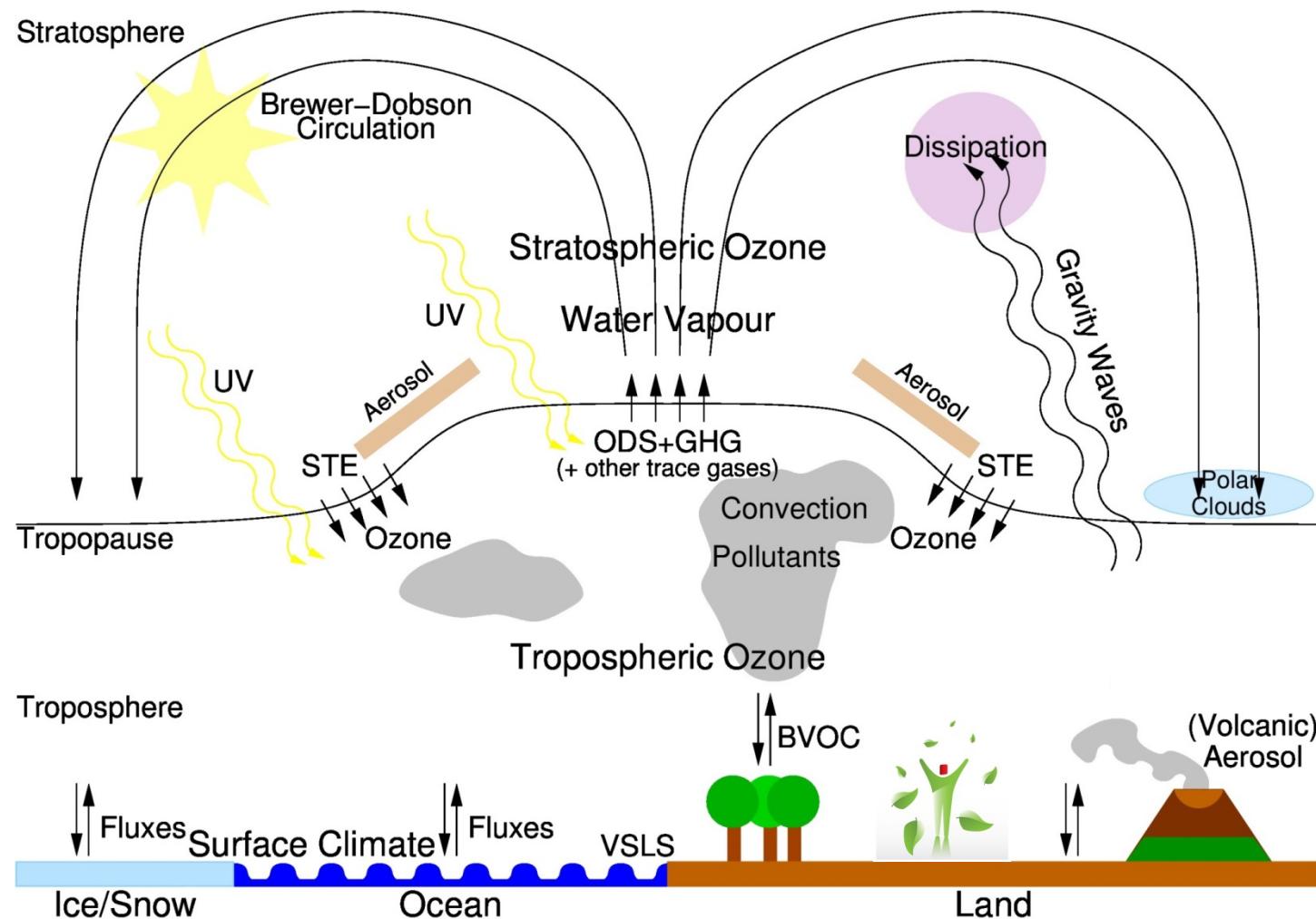
What is a Chemistry-Climate Model?

MODEL HIERARCHY

Model Hierarchy (PT4 Example)



A complex system ...



... and we are right in the middle!

Questions?

THANK YOU!

Temperature Profile

Temperature (tropical latitudes)

$$-\Gamma = \frac{dT}{dz} > -2 \frac{K}{km}$$

